

Clausur Questions for Theo.Chem. I

TC, 1a)

Given 2 functions $p_x = f \cos\varphi$ und $p_y = f \sin\varphi$. The range of angles is $\varphi \in [0, 2\pi]$. Normalize p_x to 1. What is $\langle p_x | p_y \rangle$? (Try to apply only simple arguments.) What does this result mean?

TC, 1b)

For each observable, measurable quantity x , there is a corresponding operator \mathbf{X} . E.g. if you choose, for position x , the operator $\mathbf{X} = x$; then, for momentum p , the operator is $\mathbf{P} = \dots\dots\dots$
All these operators must have some peculiar mathematical property/ies. Which one(s)?

TC, 1c)

Give the definition of a Hermitean operator A . Which property do its eigenvalues have?

TC, 2a)

Compare (I) $[\text{R}_2\text{N}-\text{CH}=\text{CH}-\text{CH}=\text{CH}-\text{CH}=\text{NR}_2]^+\text{ClO}_4^-$, (II) $[\text{R}_2\text{N}=\text{CH}-\text{CH}=\text{N}-\text{CH}=\text{CH}-\text{NR}_2]^+\text{ClO}_4^-$. How many π -pairs does I : and II : have? Which one of the two compounds absorbs light of smaller frequency?

TC, 2b)

How does the wave number $\nu = 1/\lambda$ of the first absorption band, of a conjugated π chain with equal bond lengths D and number N of atoms in the chain contributing 1 π electron each, in the VIS-UV region vary with the length $L = (N+1)D$ of the chain? Sketch a diagram. Can you derive the relation using the model of an electron in a box?

TC, 2c)

If the wave number number is $\nu = 20 \cdot 10^3$ per cm, then the wave length is $\lambda =$

The energy of the respective photons is eV.

This corresponds to an energy of kJ/mol.

TC, 3a)

An experimentalist measures the angular momentum component L_x of atomic particles along the x -axis. He obtains a statistical collection of only three different values: $\approx 1 \hbar$, or $\approx -1 \hbar$, or $\approx 0 \hbar$. Give the value of the angular momentum vector, which characterizes these particles: $l = \dots\dots\dots$,
Give the value of the total angular momentum, $|L| = \dots\dots\dots$

TC, 3b)

Which type of 2-electrons positional state function $\psi(\mathbf{r}_1, \mathbf{r}_2)$ may be combined with the 2-electrons spin direction state function $\chi(\mathbf{s}_1, \mathbf{s}_2) = \alpha(\mathbf{s}_1) \cdot \beta(\mathbf{s}_2) + \beta(\mathbf{s}_1) \cdot \alpha(\mathbf{s}_2)$? Give an explicit example for $\psi(\mathbf{r}_1, \mathbf{r}_2)$.

TC, 3c)

Does the spin function $\chi(\mathbf{s}_1, \mathbf{s}_2)$ from (3b) describe two parallel or two antiparallel spin vectors?
Draw a graphical sketch.

TC, 3d)

Explain the mechanism of covalent bonding.

Colloq, a)

Gold Nanoparticles: Typically they consist of how many atoms? Give a range N : $\dots\dots\dots$
How can one prevent the Au_N cluster particles to form bigger colloidal particles?

Colloq, b)

Sketch the setup to measure the electric conductivity of single molecules.
Which molecules, for instance, are interesting in this context?

Colloq, c)

How does the fluorescence spectrum of a semiconductor nanoparticle look like?
Of a single quantum dot? Of a surface covered with many of them?

Surface, a)

You fill a plastic box with a hole at the bottom with water until the water starts to run out of the hole. This hole has a radius of 0.1 mm. As you know the height of the water level is 14.7 cm. You add a substance which dissolves in the water. The final concentration is 100 mMol/l. You observe that the water starts to run out of the box again until it reaches a height of 12 cm. The temperature is 25°C. - What is the surface tension after adding the substance?

Surface, b)

Please estimate the surface excess in molecules per nm².

Surface, c)

Instead of adding the substance, you heat the water to 40°C. Does water run out (starting from a height of 14.7 cm)? If yes, to which level does it run out?