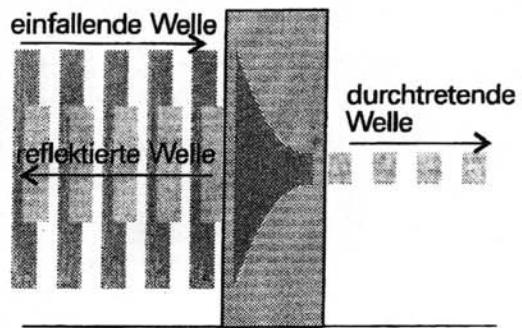
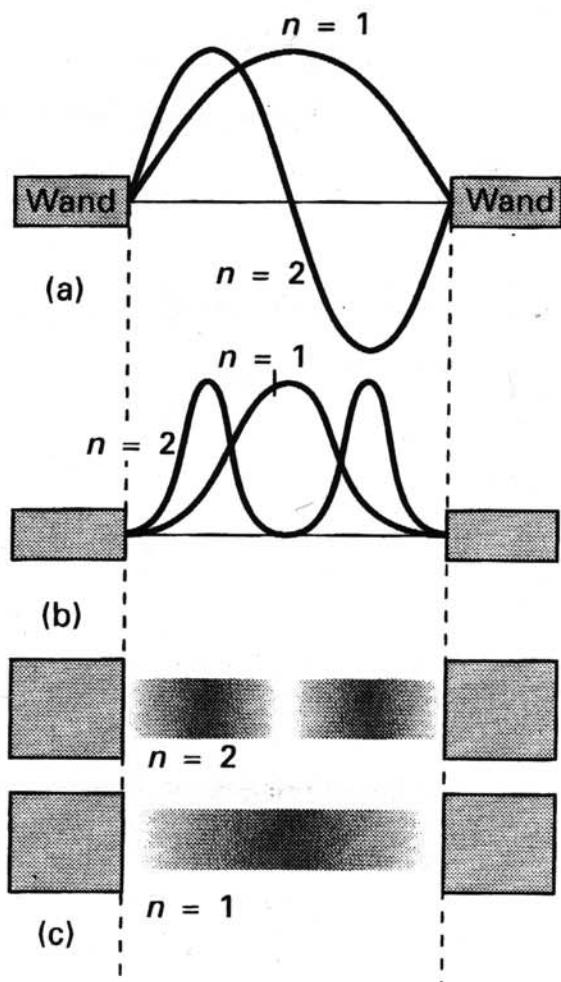


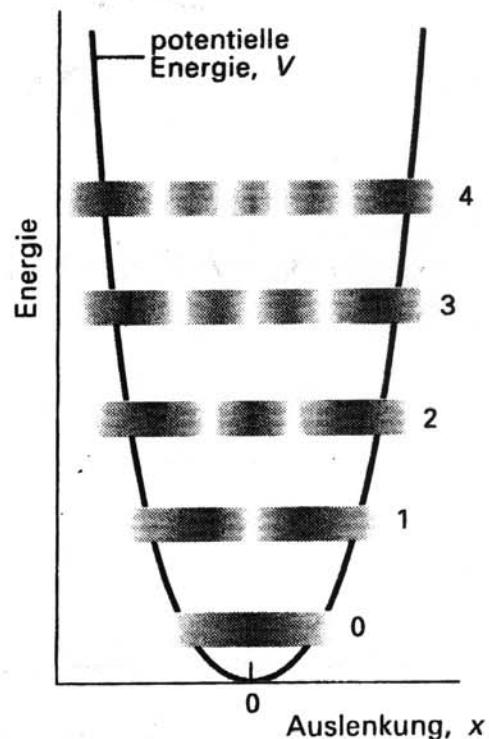
**Fig. 11.10** Interferences of electron waves



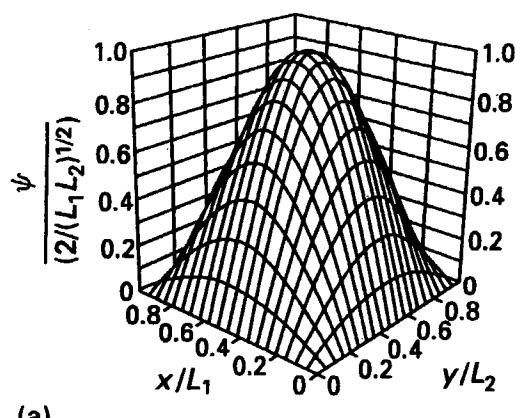
**Fig. 12.10** Tunneling and reflection of particles at a barrier



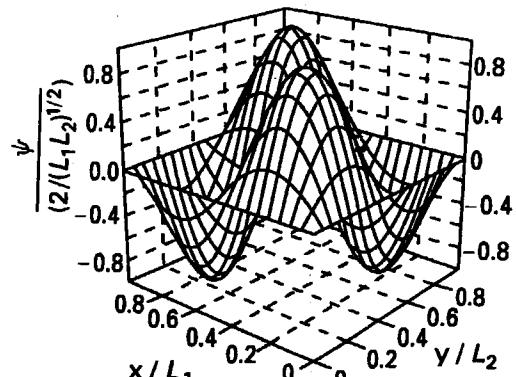
**Fig. 12.4** Two lowest states of electron in box



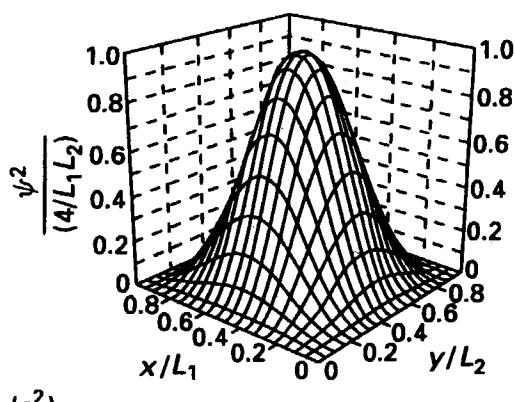
**Fig. 12.13** Energies and distributions of a harmonic oscillator (diatomic molecule)



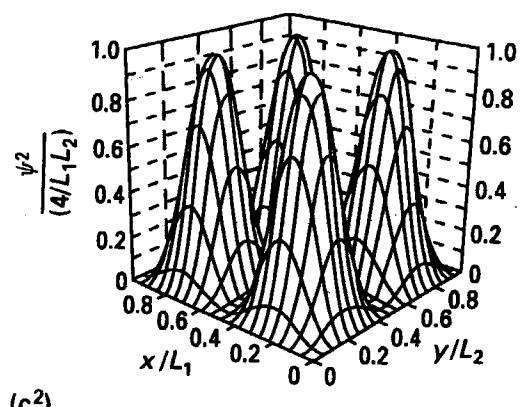
(a)



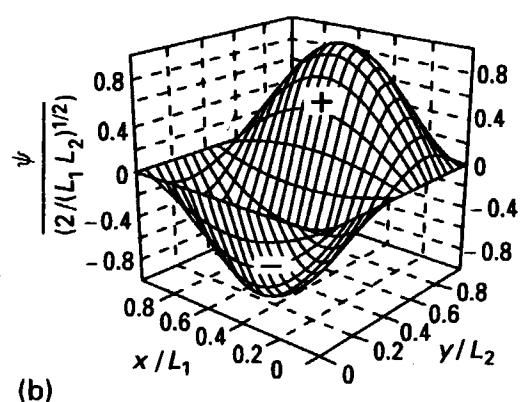
(c)



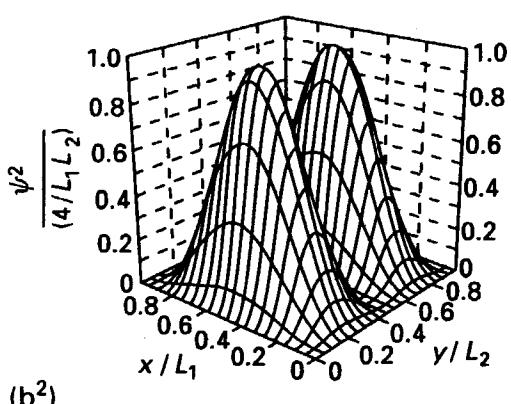
$(a^2)$



$(c^2)$

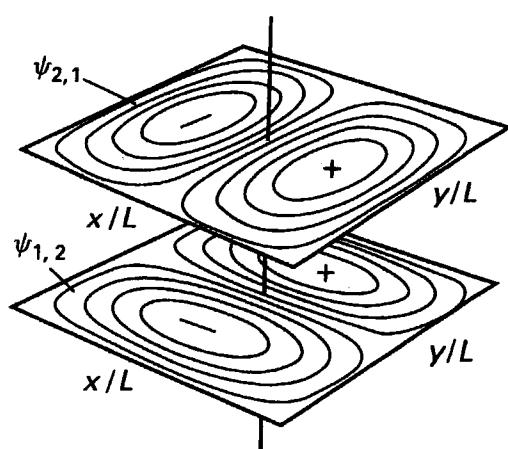


(b)

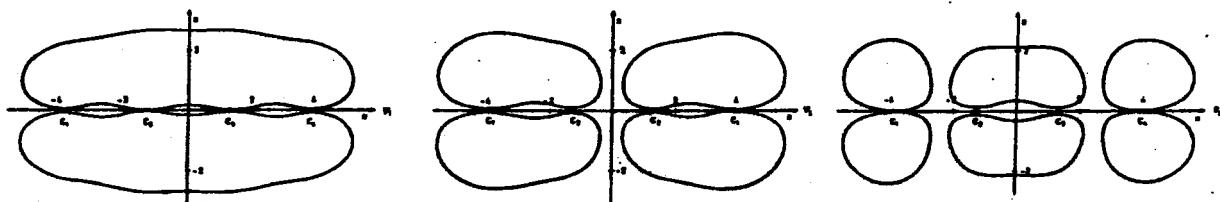
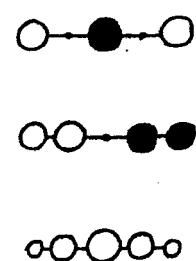
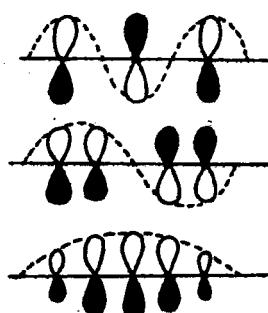
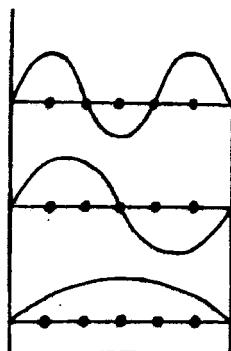


$(b^2)$

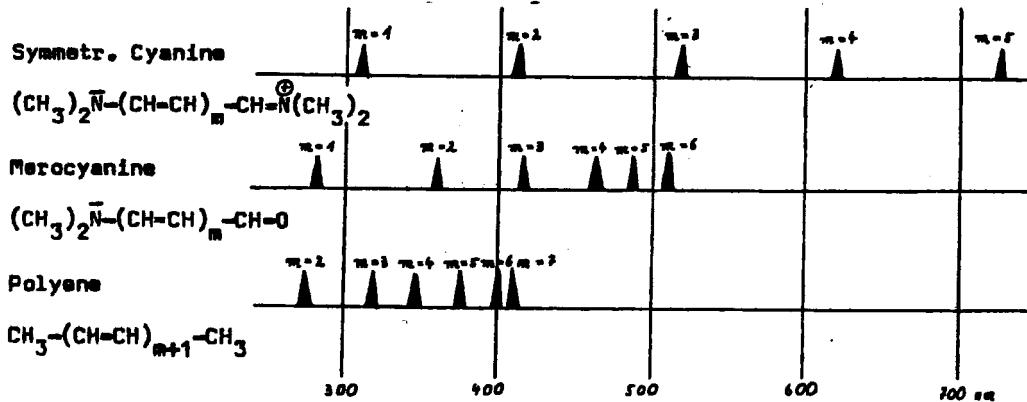
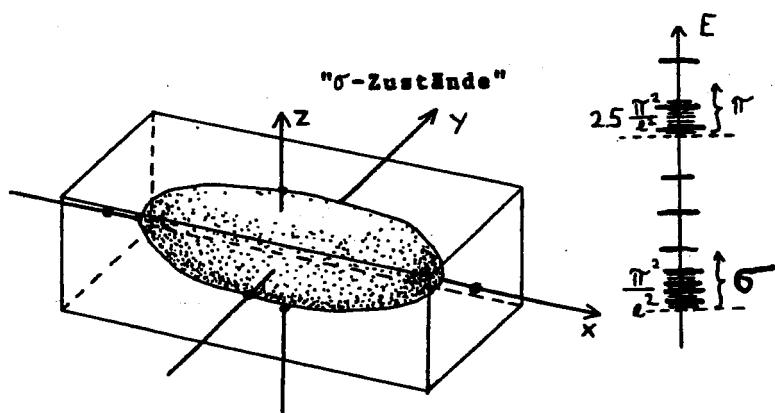
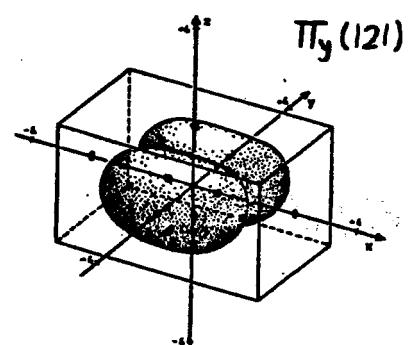
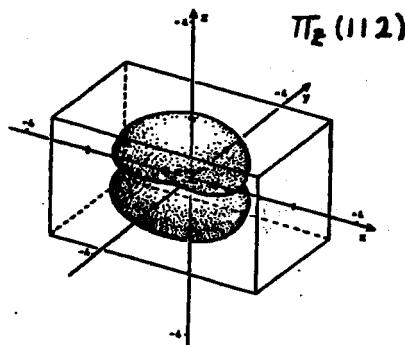
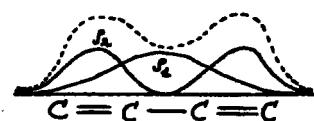
**Fig. 12.7** Three different states  $(1,1)$ ,  $(1,2)$ ,  $(2,2)$  of an electron in a rectangular box  $[L_1, L_2]$



**Fig. 12.8** So-called contour plot of states  $(1,2)$  and  $(2,1)$

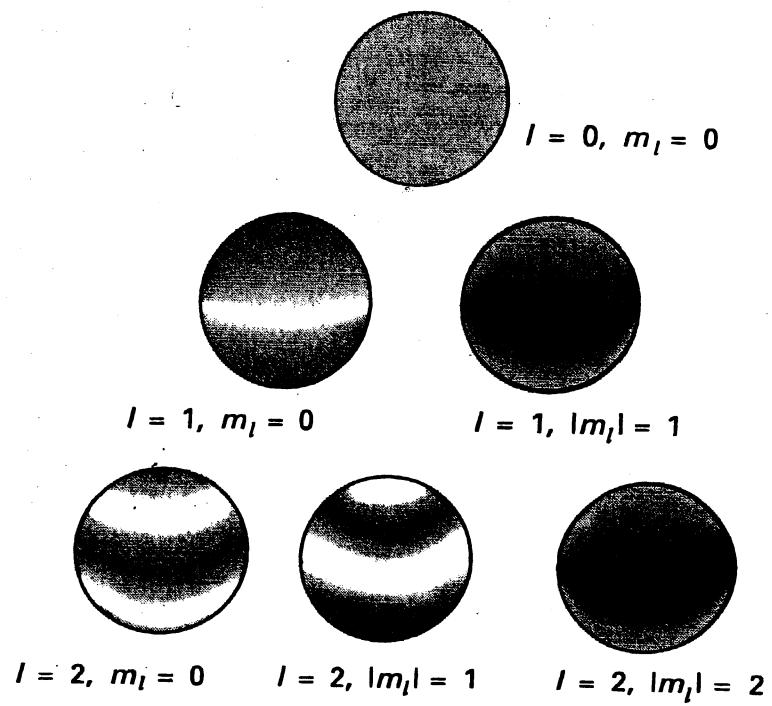


Elektronendichte im Butadien:

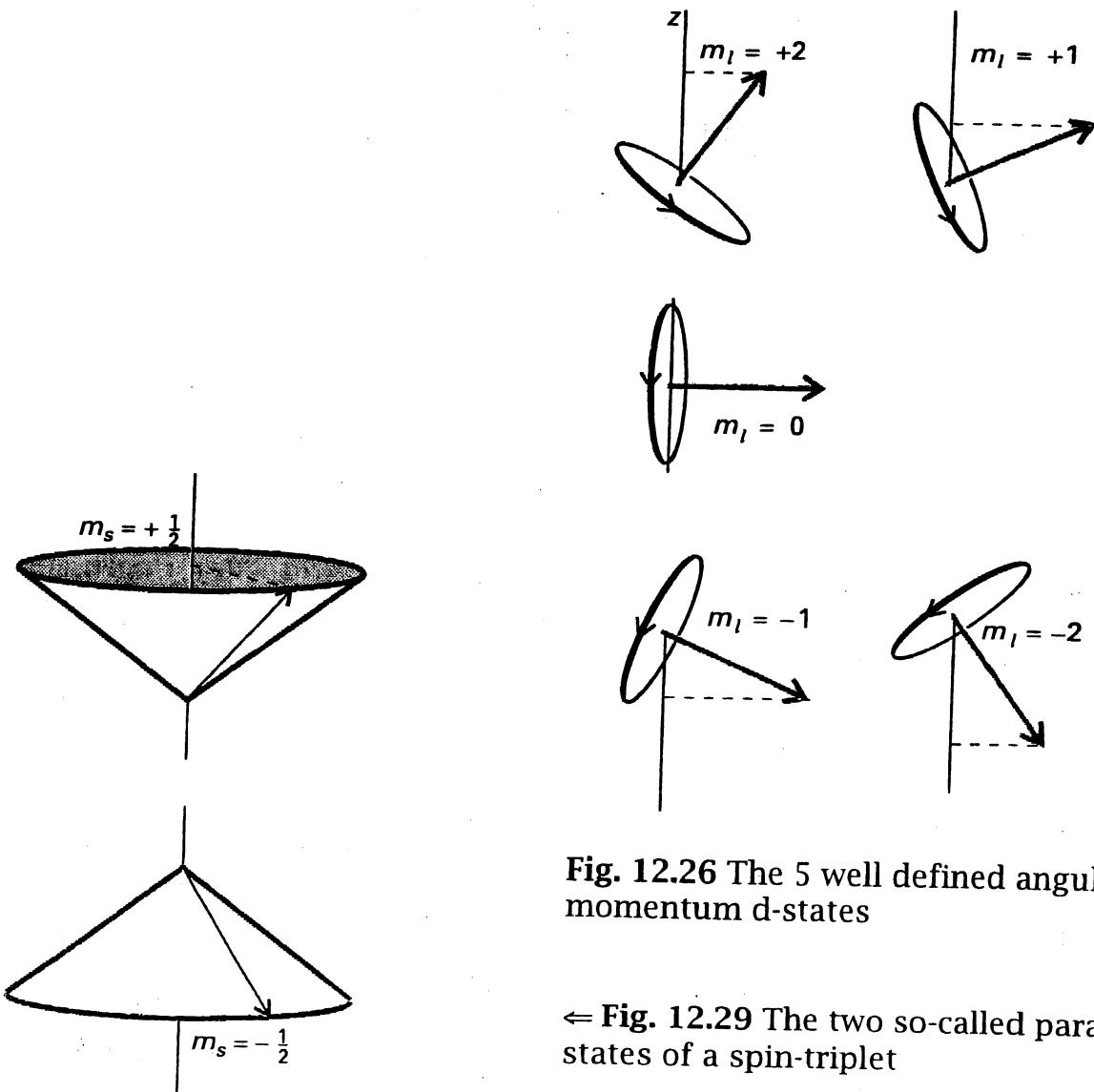


**Tab. 12.3** Spherical harmonics

$l$	$m_l$	$Y_{l,m}$	$Z_{l,m}$
0	0	$(\frac{1}{4\pi})^{1/2}$	1
1	0	$(\frac{3}{4\pi})^{1/2} \cos\theta$	$\frac{z}{r}$
	$\pm 1$	$\pm(\frac{3}{8\pi})^{1/2} \sin\theta e^{\pm i\phi}$	$\frac{x; y}{r}$
2	0	$\pm(\frac{5}{16\pi})^{1/2} (3\cos^2\theta - 1)$	$\frac{2z^2 - x^2 - y^2}{r^2}$
	$\pm 1$	$\pm(\frac{15}{8\pi})^{1/2} \cos\theta \sin\theta e^{\pm i\phi}$	$\frac{xz; yz}{r^2}$
	$\pm 2$	$\pm(\frac{15}{8\pi})^{1/2} \cos\theta \sin\theta e^{\pm i\phi}$	$\frac{x^2 - y^2; xy}{r^2}$



**Fig. 12.25** Angular distribution of particles in s, p, d-states



**Fig. 12.26** The 5 well defined angular momentum d-states

← **Fig. 12.29** The two so-called parallel states of a spin-triplet